

# Experiments with a high visibility lithium atom interferometer

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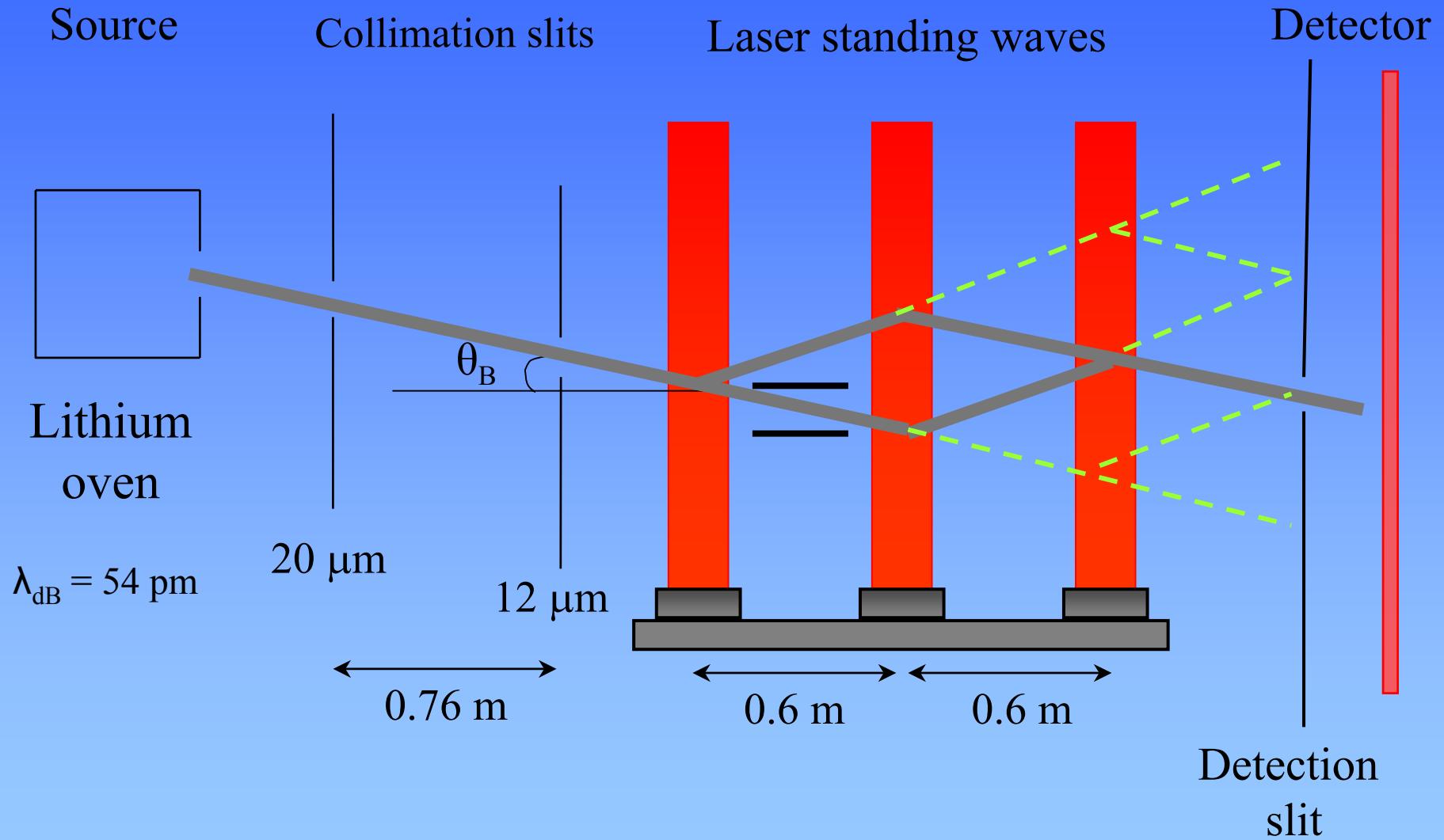
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TOULOUSE – FRANCE  
*Les Houches – February 2005*



# Plan

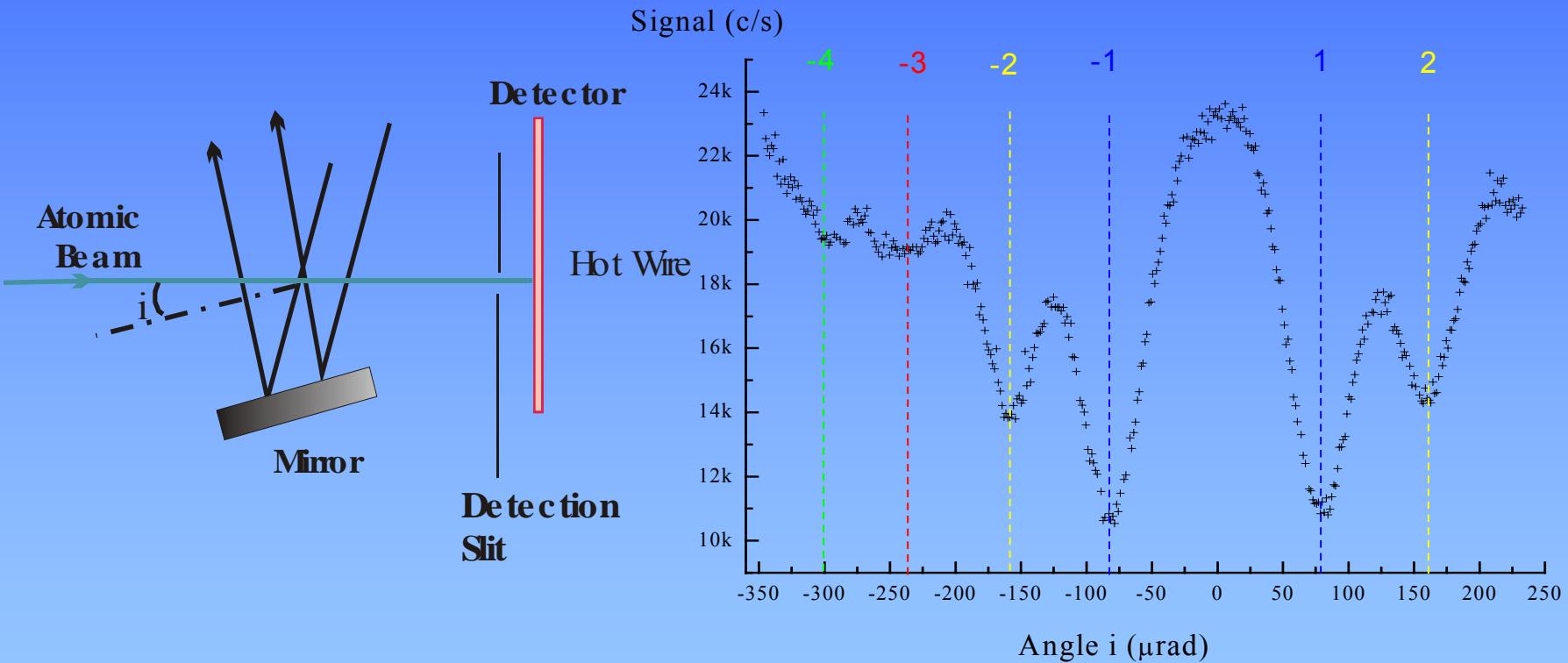
- I. Our atom interferometer
- II. Experiments with a magnetic field gradient
- III. Lithium atom polarizability measurement

# Experimental set-up



# Diffraction of the lithium atomic wave

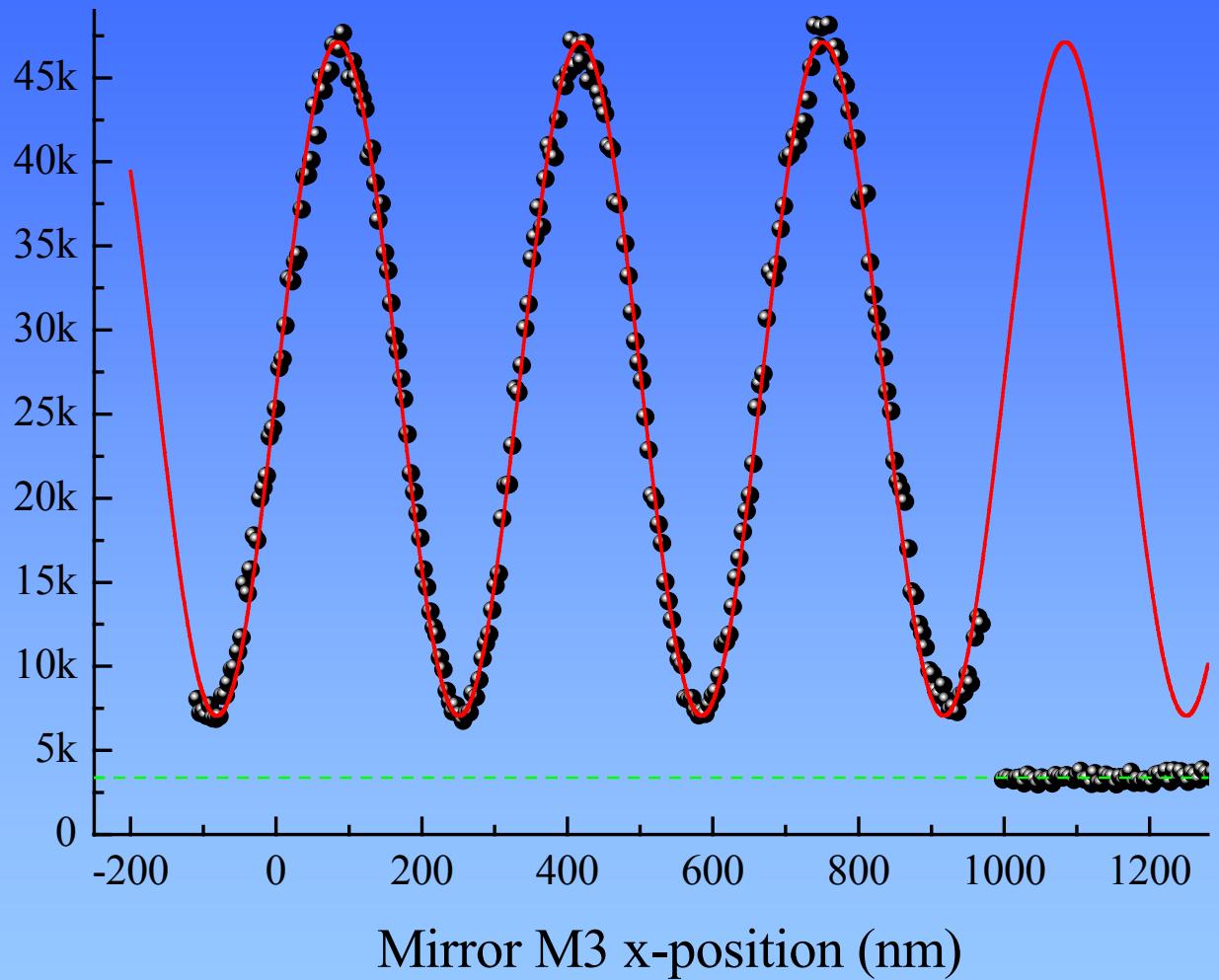
As a function of the orientation of the mirror used for the laser standing wave, the Bragg condition is fulfilled for various orders  $p$ , here from  $p = -2$  to  $p = +4$



Laser standing waves :  $\delta_L/(2\pi) = 1.2 \text{ GHz}$ ,  $w_0 = 3.1 \text{ mm}$ ,  $P_L = 240 \text{ mW}$

# Lithium atom interference fringes

Signal (c/s)



$$p = 1$$

Fringe visibility  $V = 84.5 \pm 1\%$

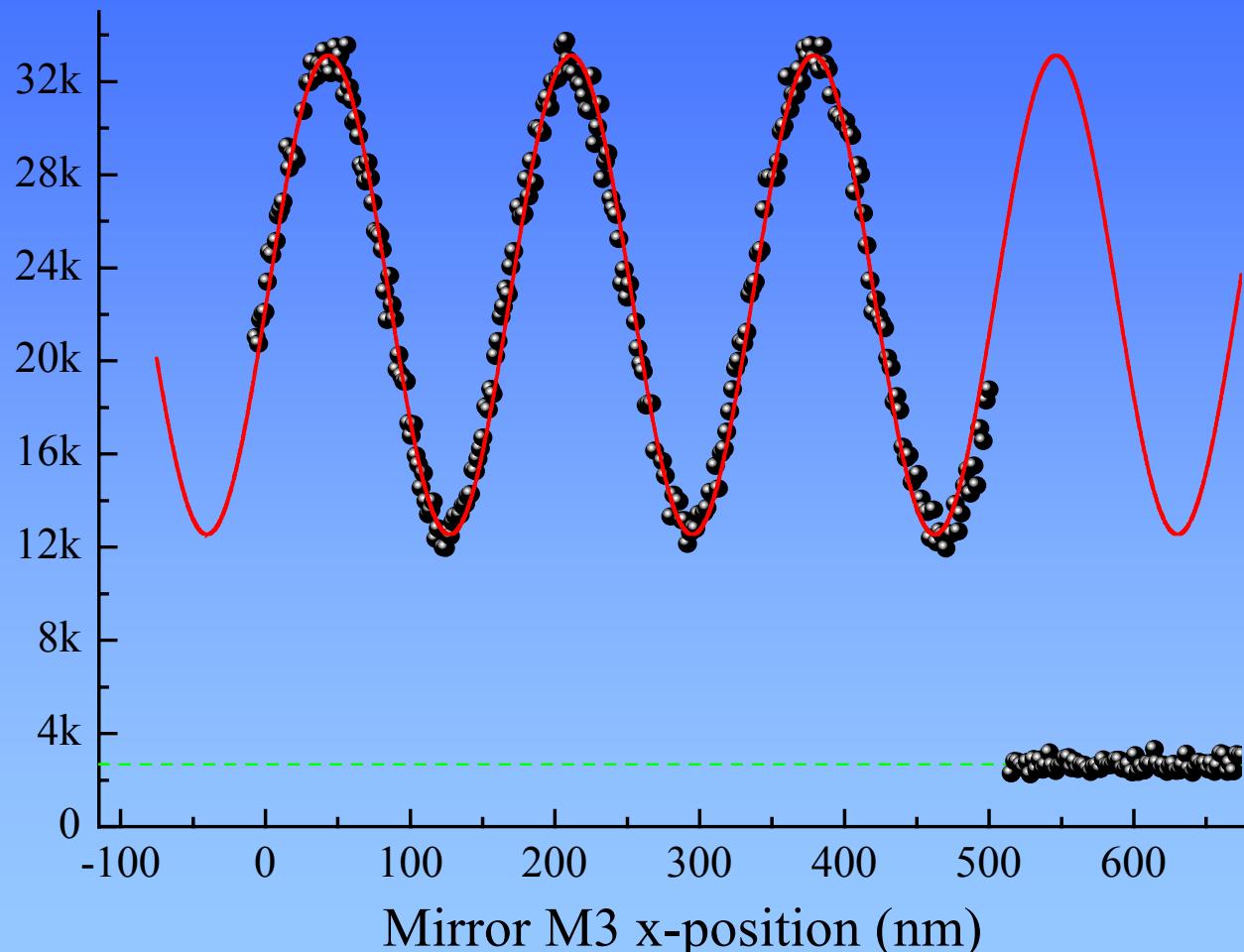
Output Flux  
 $I_0 = 23700 \text{ c/s}$

Phase sensitivity  
 $25 \text{ mrad} / \sqrt{\text{Hz}}$

Laser standing waves:  $\delta/(2\pi) = 2.8 \text{ GHz}$ ,  $w_0 = 5.0 \text{ mm}$ ,  $P = 150 \text{ mW}$

# Atom fringes using Bragg order 2 diffraction

Signal (c/s)



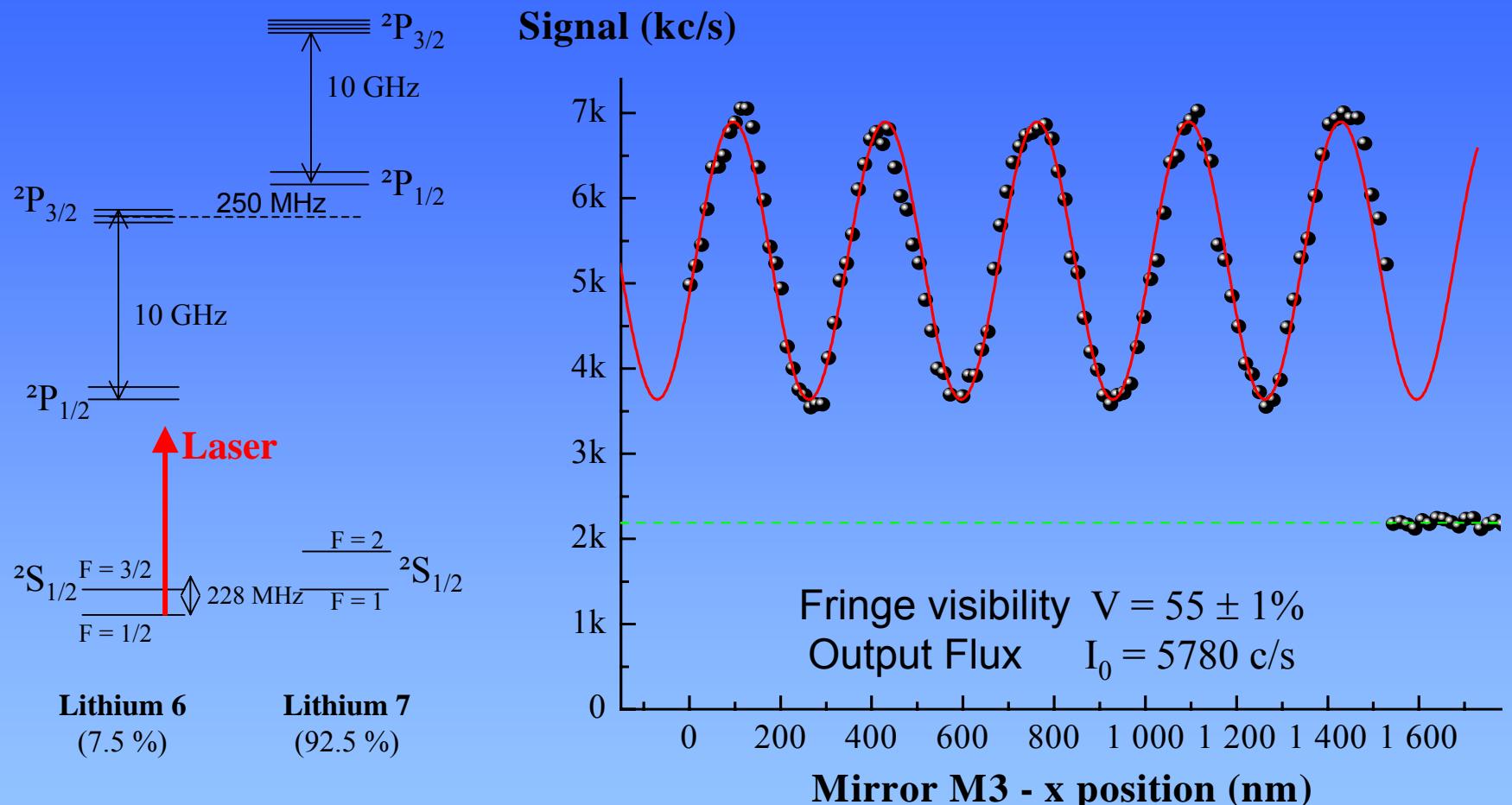
$p = 2$

Fringe visibility  
 $V = 54 \pm 1\%$

Output Flux  
 $I_0 = 20180 \text{ c/s}$

Laser standing waves:  $\delta/(2\pi) = 3.1 \text{ GHz}$ ,  $w_0 = 1.8 \text{ mm}$ ,  $P = 460 \text{ mW}$

# Isotope 6 interference fringes



## II. Experiments with a magnetic field gradient

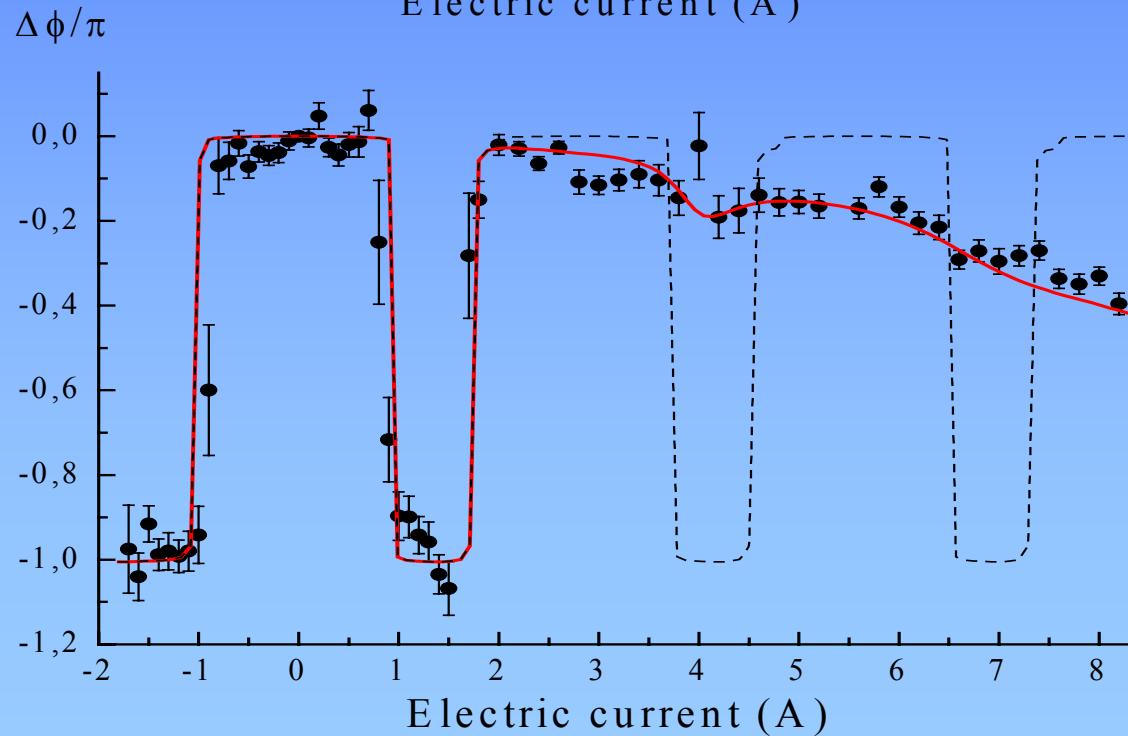
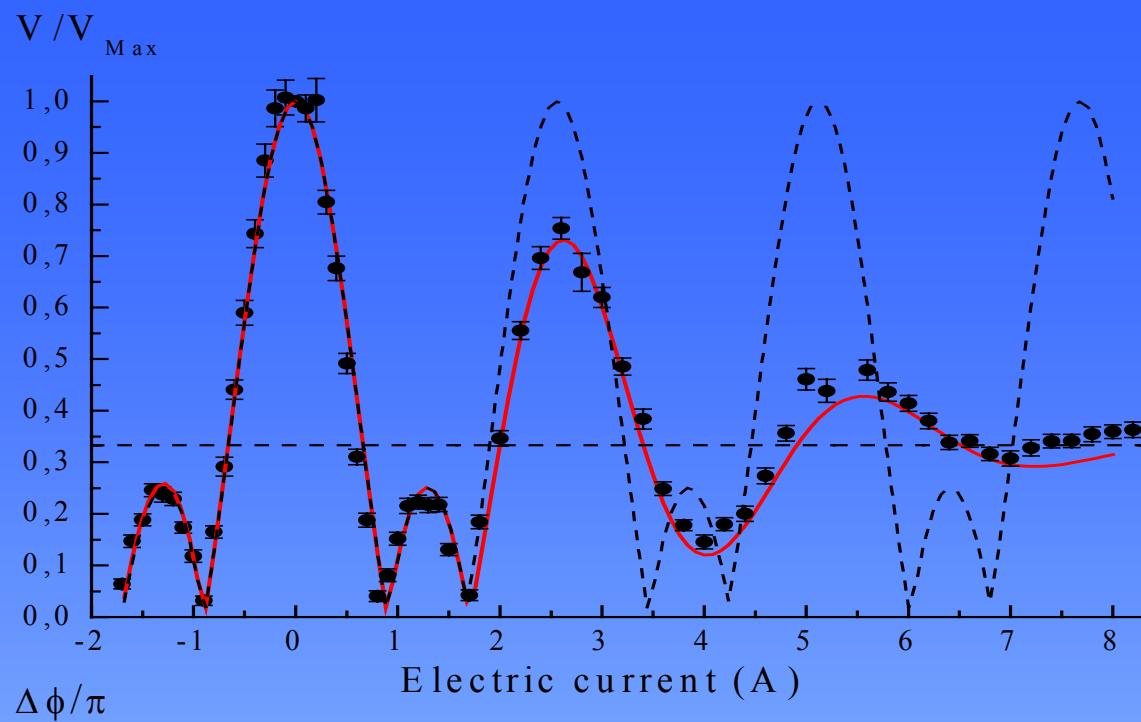
- The  $|F, M_F\rangle$  sublevels are the eigenstates
- Adiabatic approximation
  - The magnetic field  $B$  does not vanish or rotate too rapidly
- Zeeman Phase Shift

$$\Delta\phi(F, M_F) = \varphi M_F$$

$$\varphi = \frac{g_F \mu_B}{\hbar v} \int \frac{dB(s)}{dx} \Delta x(s) ds$$

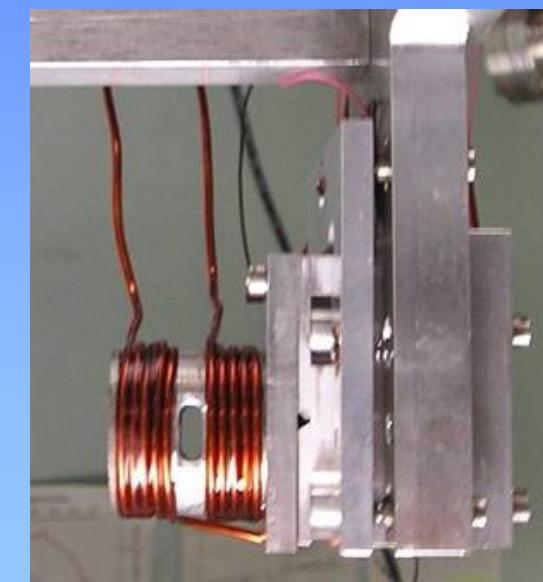
- For  ${}^7\text{Li}$  ( $I = 3/2$ ),  ${}^2\text{S}_{1/2}$ ,  $F=1$  and  $F=2 \rightarrow 8$  sublevels (Landé factors  $g_F = \pm 1/2$ )  
Optical pumping in  ${}^2\text{S}_{1/2, F=1}$

$$\frac{V}{V_{MAX}} = \frac{1 + 2 \cos(\varphi)}{3}$$



----- Theoretical simulation  
without velocity dispersion

—— Theoretical simulation  
including velocity dispersion



# Magnetic rephasing experiment with ${}^6\text{Li}$ ( $I = 1$ )

$V/V_{MAX}$

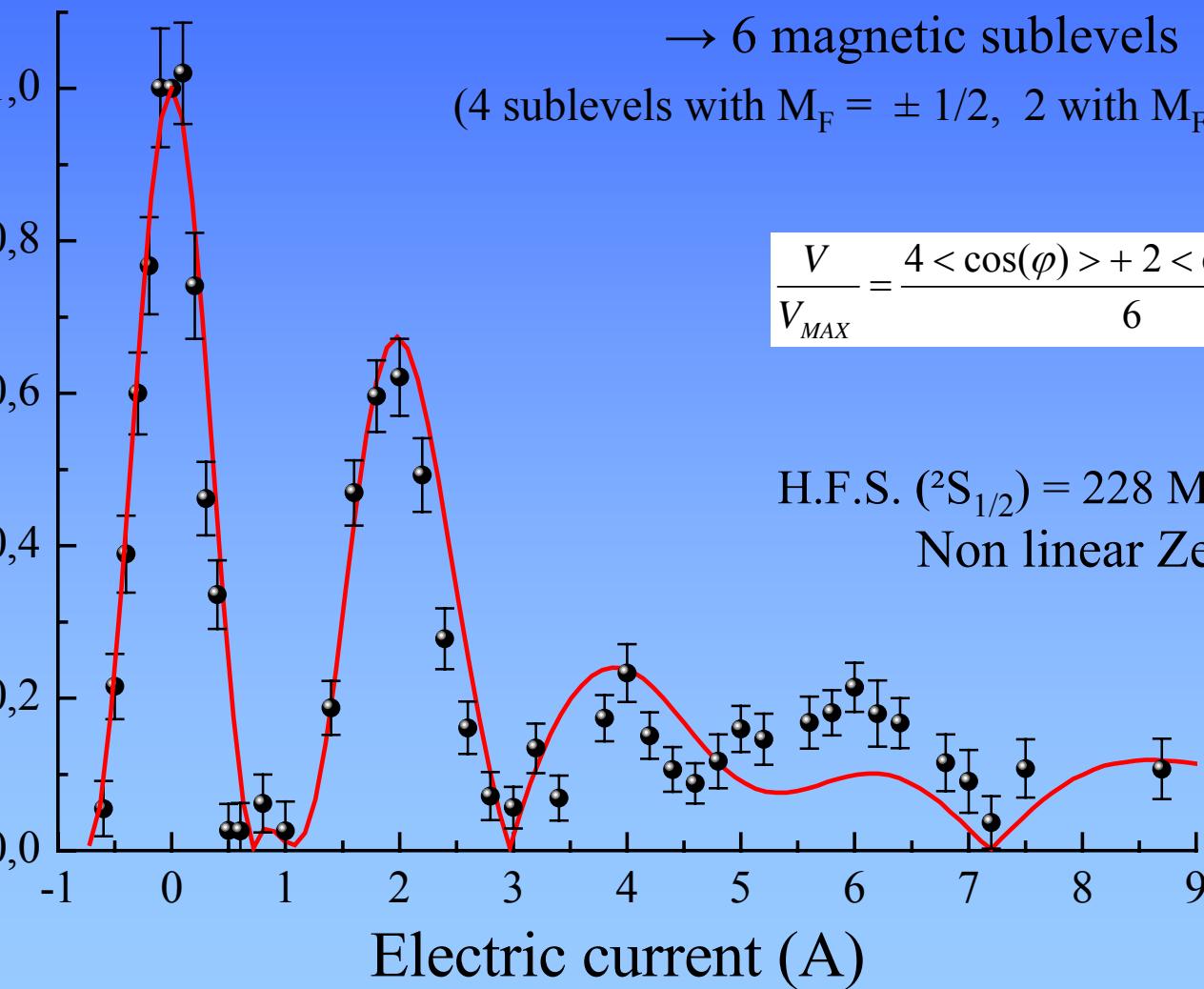
${}^2\text{S}_{1/2}$ ,  $F=1/2$  and  $F=3/2$  ( $g_F = \pm 1/3$ )

→ 6 magnetic sublevels

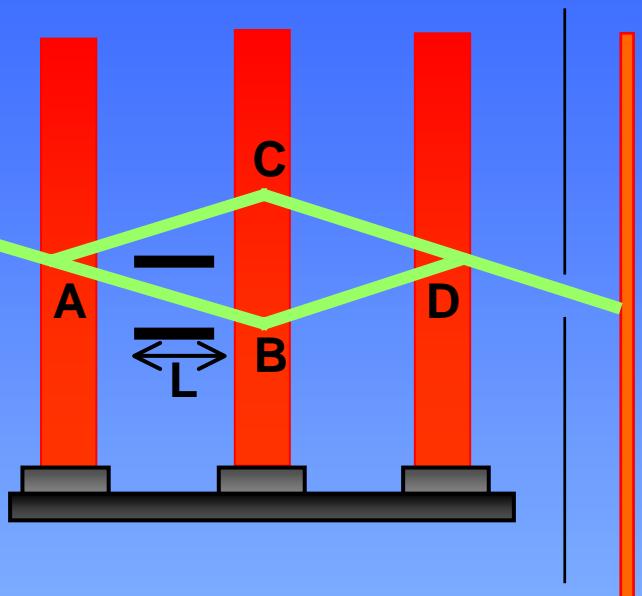
(4 sublevels with  $M_F = \pm 1/2$ , 2 with  $M_F = \pm 3/2$ )

$$\frac{V}{V_{MAX}} = \frac{4 < \cos(\varphi) > + 2 < \cos(2\varphi) >}{6}$$

H.F.S. ( ${}^2\text{S}_{1/2}$ ) = 228 MHz smaller than for  ${}^7\text{Li}$   
Non linear Zeeman effects !



### III. Lithium atom polarizability measurement



$$\Delta\Phi = \int_{ABD} U(t) dt / \eta - \int_{ACD} U(t) dt / \eta$$

$$U = \text{Perturbation} = -4\pi\epsilon_0\alpha E^2/2$$

D.E. Pritchard et al, Phys. Rev. A **51**, 3883  
Electric polarizability of sodium atom

Stark Phase Shift

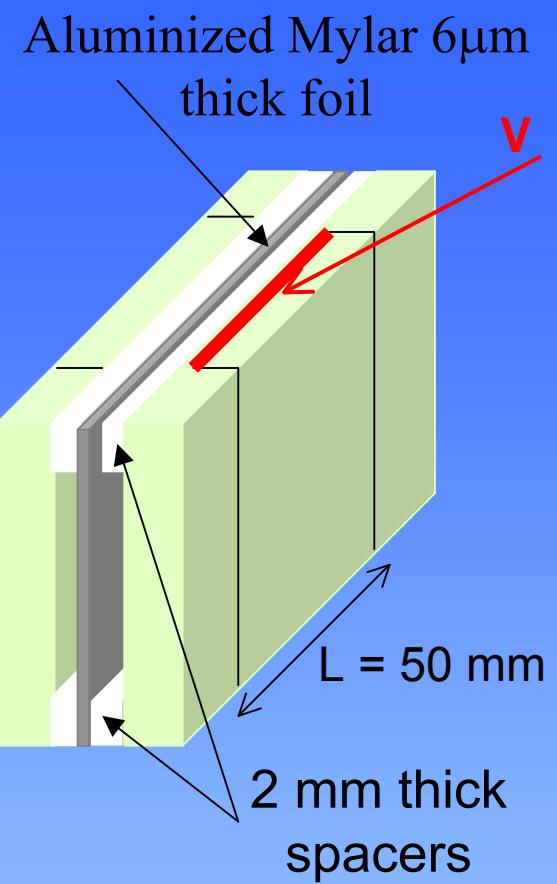
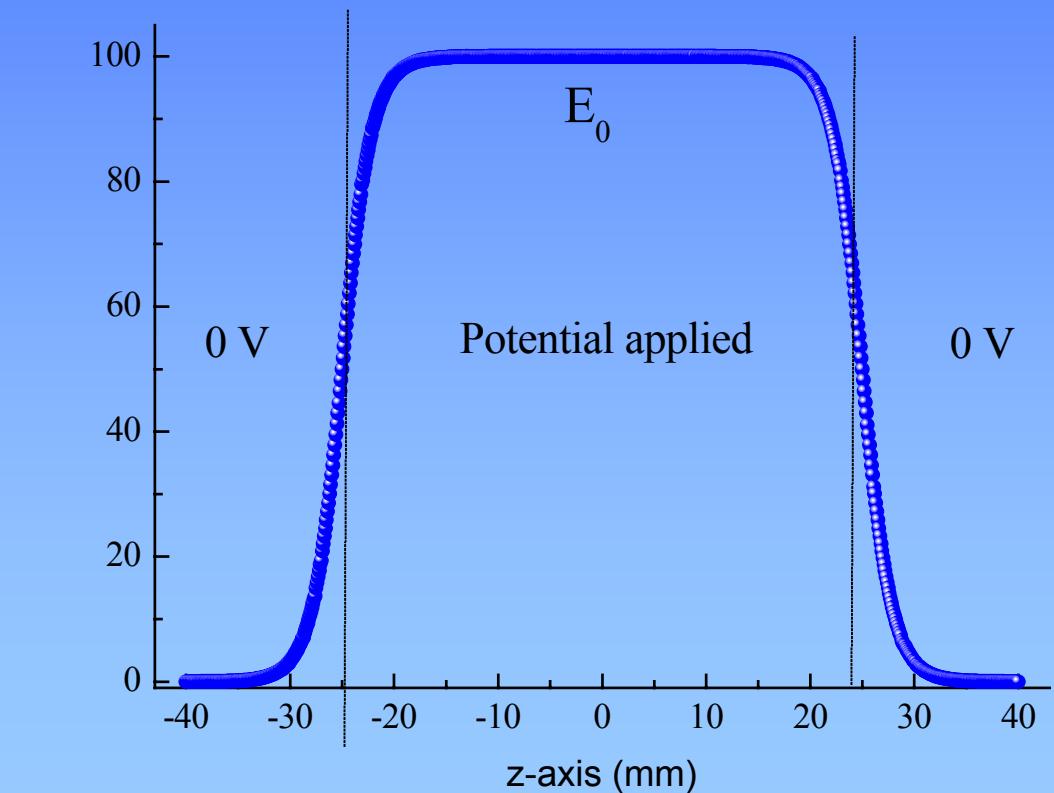
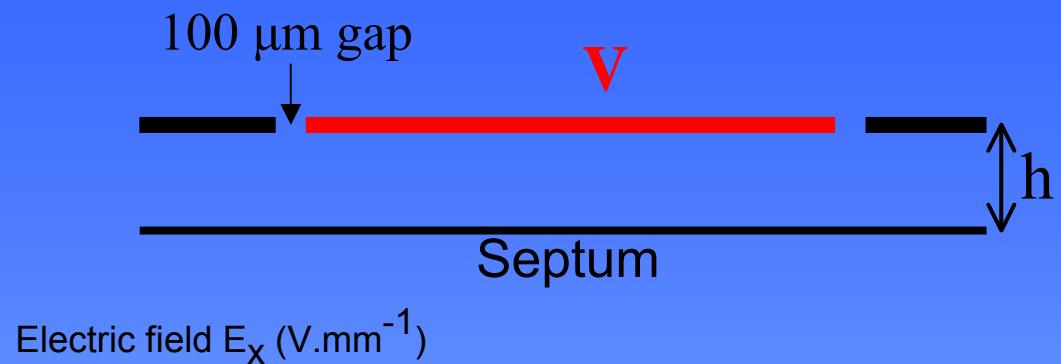
$$\Delta\phi_{\text{Stark}} = 4\pi\epsilon_0\alpha \frac{E^2 L}{2\eta u}$$

proportional to  $E^2/u$   
( $u$  = mean beam velocity)

Two main difficulties :

1. Capacitor geometry, beam separation of 100 micrometers for  $p = 1$
2. Precise knowledge of the mean beam velocity of the atomic beam

# The capacitor built in Toulouse



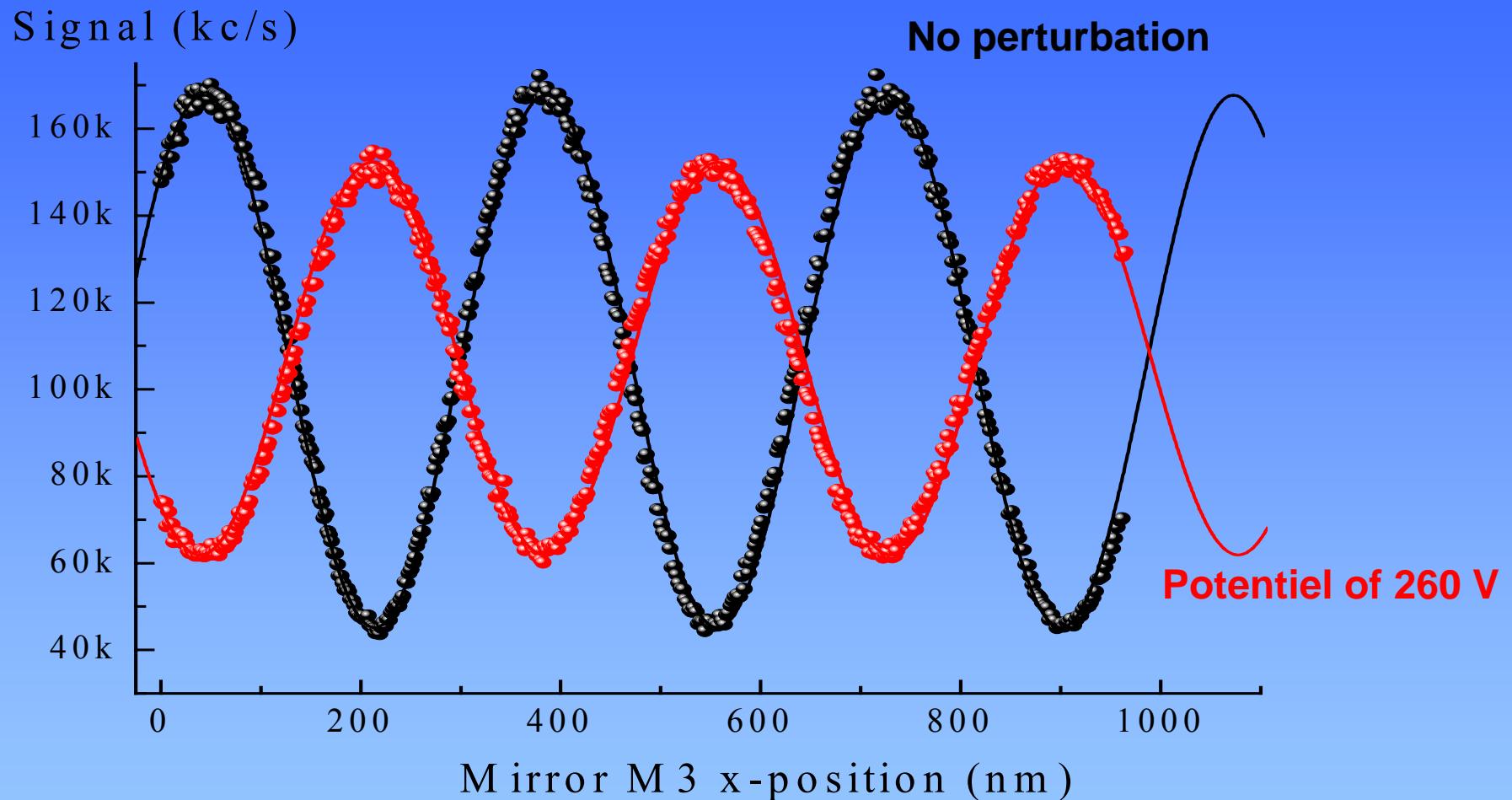
$$L_{\text{eff}} = \int E^2(z) dz / E_0^2$$

$$L_{\text{eff}} = 46,82 \pm 0,10 \text{ mm}$$

# Lithium atom polarizability

Error source	Systematic errors	
Effective length	$L_{\text{eff}} = 46,82 \pm 0,10 \text{ mm}$	0.21 %
Electrode spacing	$h = 2,027 \pm 0,002 \text{ mm}$	$2 \times 0.1 \%$
Contact potentials		0.01 %

# Observation of the phase Stark shift



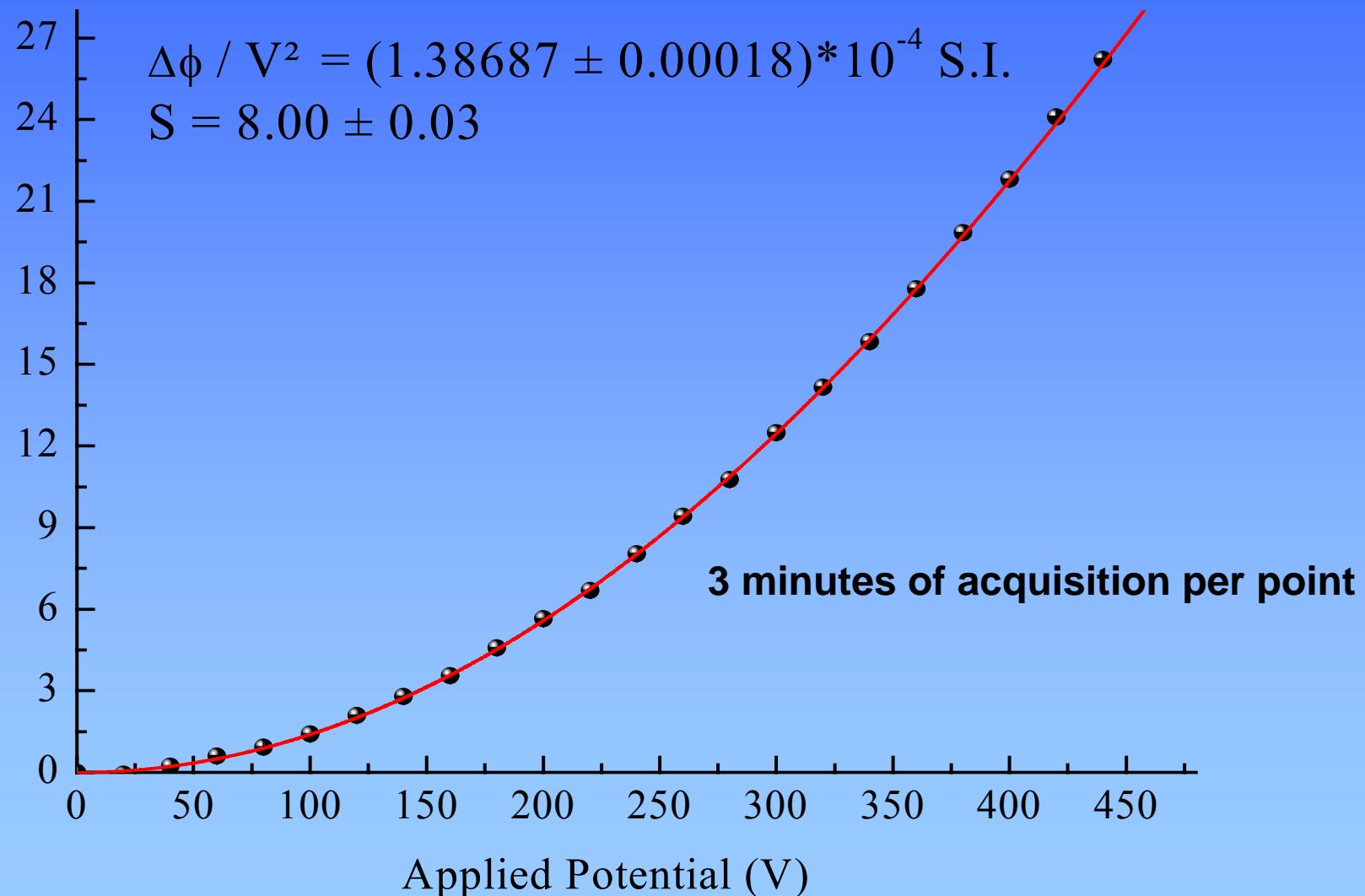
$$I_o = 105\,000 \text{ c/s}$$
$$V_0 = 60 \%$$

$$\Delta\Phi_{\text{Stark}}(260 \text{ V}) \sim 3\pi \text{ rad}$$

# Measurement of the phase shift

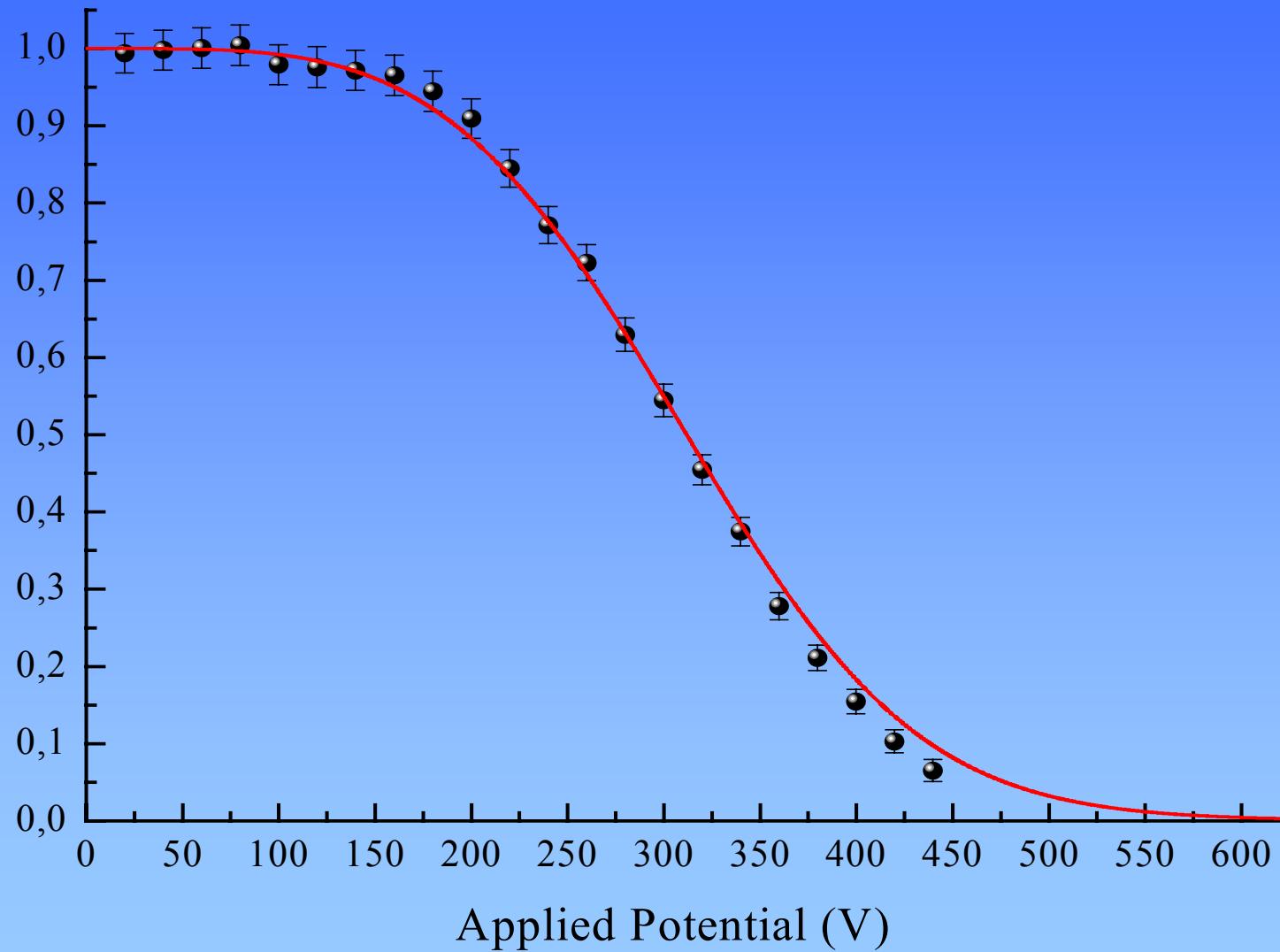
(Small corrections due to the velocity dispersion are included)

$\Delta\phi$  (rad)



# Fringe visibility as a function of the applied voltage

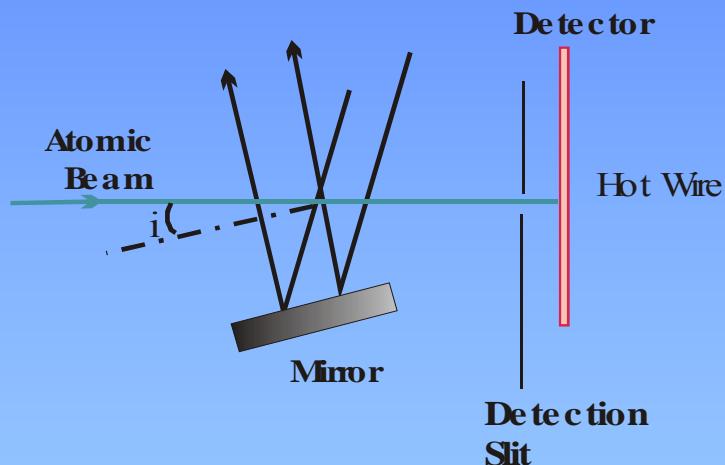
Fringe Visibility (normalized)



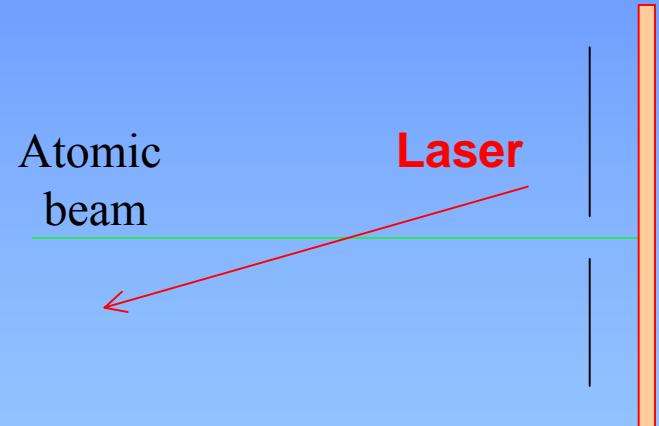
# Mean velocity of the atomic beam

Two main methods

Bragg Diffraction



Doppler effect



$$u = 1075.7 \pm 5.4 \text{ m/s}$$

$$u = 1066.1 \pm 8.1 \text{ m/s}$$

# Lithium atom polarizability

Error source	Systematic errors	Statistical errors
Effective length	0.21 %	
Electrode spacing	$2 \times 0.1 \%$	
Contact potentials	0.01 %	
Phase measurement		$\Delta\Phi / V^2 = (1.38687 \pm 0.00018) \times 10^{-4}$ SI 0.1 %, analysis in progress
Velocity measurement		$u = 1073.1 \pm 7.0$ m/s      0.65 %

$$\alpha(\text{Li}) = 24,74(8)_{\text{systematic}}(16)_{\text{statistical}} \times 10^{-30} \text{ m}^3$$

# Conclusions

## Atom interference fringes

High visibility and high flux,  
many possibilities (order 2, isotope 6)

## External fields as tools

- Internal structure of the atom, nuclear spin
- Lithium atom polarizability measurement

$$\alpha(\text{Li}) = 24,74(8)_{\text{systematic}}(16)_{\text{statistical}} \times 10^{-30} \text{ m}^3$$

Our measurement ( $\sim 1\%$  accuracy) compares very well to the latest done by Molof et al. in 1974  $\alpha = (24,3 \pm 0,5) \times 10^{-30} \text{ m}^3$  (precision of 2 %)

*A. Molof et al., Phys.Rev. A, 10, (1331-1140)*



# D. E Pritchard et al (1995) : Sodium atom electric polarizability

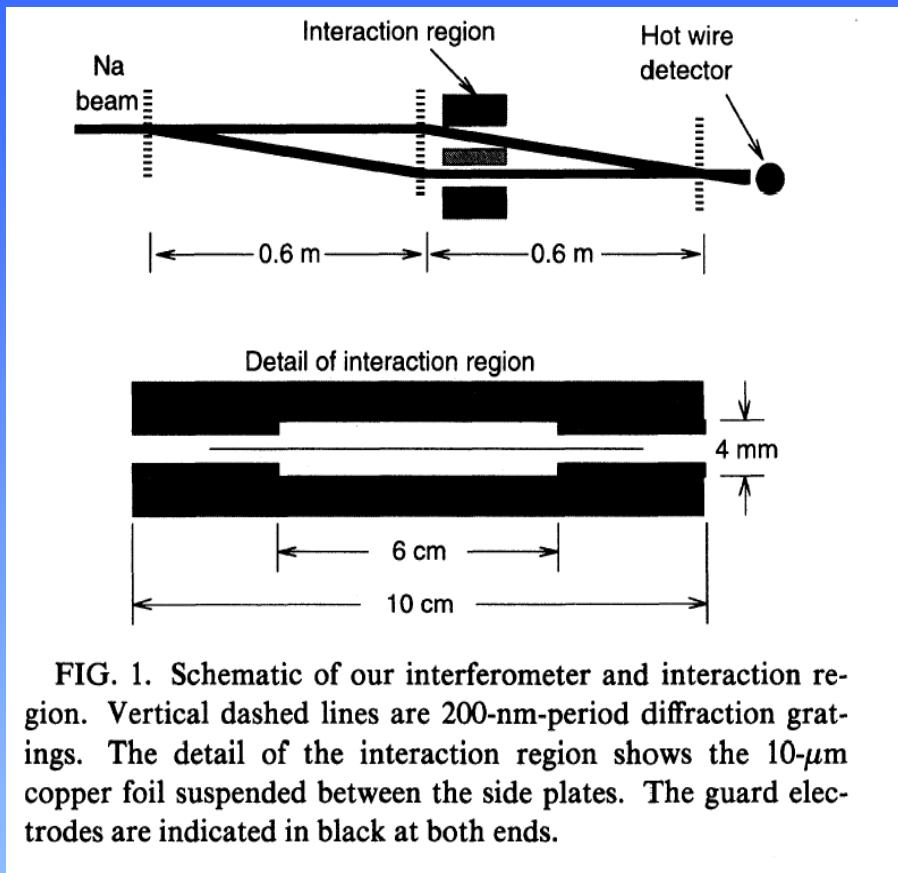


FIG. 1. Schematic of our interferometer and interaction region. Vertical dashed lines are 200-nm-period diffraction gratings. The detail of the interaction region shows the 10- $\mu\text{m}$  copper foil suspended between the side plates. The guard electrodes are indicated in black at both ends.

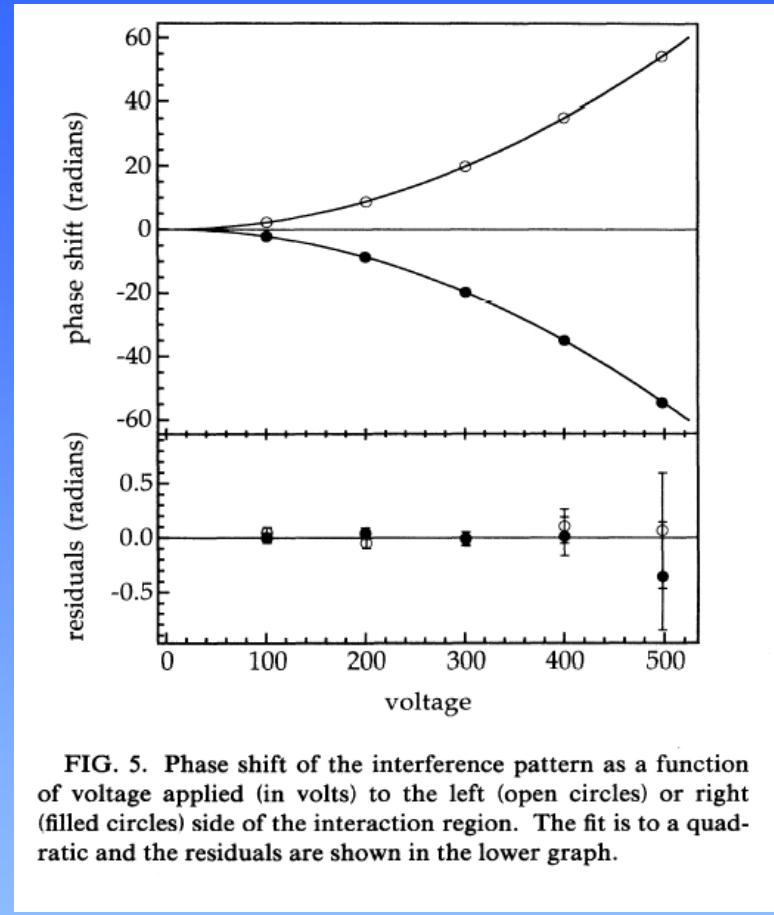


FIG. 5. Phase shift of the interference pattern as a function of voltage applied (in volts) to the left (open circles) or right (filled circles) side of the interaction region. The fit is to a quadratic and the residuals are shown in the lower graph.

$$U = -4\pi\epsilon_0\alpha E^2/2 \rightarrow \Delta\Phi \text{ proportional to } E^2/v$$

Accuracy on  $\alpha$  :  $\pm 0.25\%$  statistical  $\pm 0.25\%$  systematics

Main difficulties: capacitor geometry,  
precise knowledge of velocity distribution

# Index of refraction of gases for atomic waves

The index  $n$  is complex and expresses the effect of collisions.  
 $(n-1)$  is proportional to the gas density  
 $\text{Re}(n-1)$  measures the dephasing of the wave  
 $\text{Im}(n-1)$  measures the attenuation of the wave

Experiments with sodium by D. Pritchard et al.

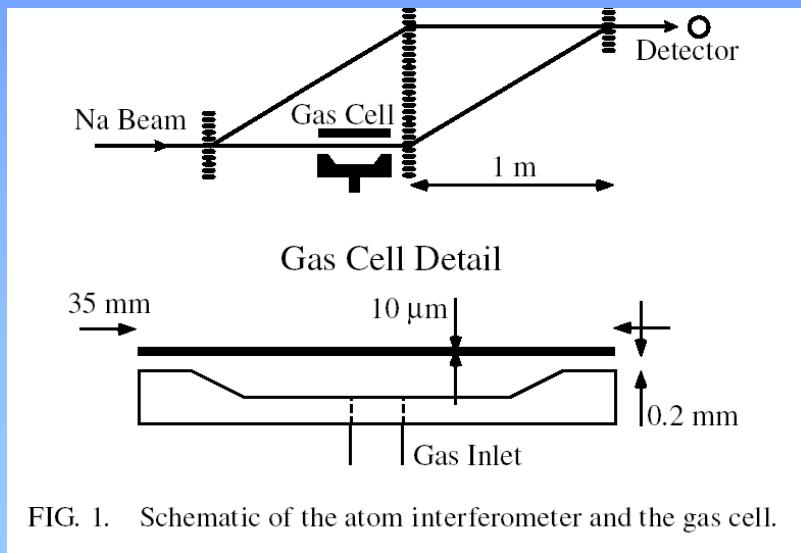


FIG. 1. Schematic of the atom interferometer and the gas cell.

$$\rho(v) = \frac{\text{Re}(n-1)}{\text{Im}(n-1)}$$

